

WEEKLY TEST TYJ -1 TEST - 15 R & B SOLUTION Date 04-08-2019

[PHYSICS]

1. (c)
$$\frac{A}{B} = \frac{\text{Force}}{\text{Force}} = [M^0 L^0 T^0]$$

$$Ct = \text{angle} \implies C = \frac{\text{Angle}}{\text{Time}} = \frac{1}{T} = T^{-1}$$

$$Dx = \text{angle} \implies D = \frac{\text{Angle}}{\text{Distance}} = \frac{1}{L} = L^{-1}$$

$$\therefore \frac{C}{D} = \frac{T^{-1}}{L^{-1}} = [M^0 L T^{-1}]$$

 (d) Maximum error in measuring mass = 0.001 g, because least count is 0.001 g. Similarly, maximum error in measuring volume is 0.01 cm³.

$$\frac{\Delta \rho}{\rho} = \frac{\Delta M}{M} + \frac{\Delta V}{V} = \frac{0.001}{20.000} + \frac{0.01}{10.00}$$
$$= (5 \times 10^{-5}) + (1 \times 10^{-3}) = 1.05 \times 10^{-3}$$
$$\Delta \rho = (1.05 \times 10^{-3}) \times \rho$$
$$= 1.05 \times 10^{-3} \times \frac{20.000}{10.00} = 0.002 \text{ g cm}^{-3}$$

3. (d)
$$\frac{C^2}{g} = \frac{L^2 T^{-2}}{L T^{-2}} = [L]$$

4. (c) Graphically, the area of v-t curve represents displacement

$$S = \frac{1}{2} v_{\text{max}} t_1 \quad \text{or} \quad t_1 = \frac{2S}{v_{\text{max}}}$$

$$2S = v_{\text{max}}t_2 \quad \text{or} \quad t_2 = \frac{2S}{v_{\text{max}}}$$

$$5S = \frac{1}{2}v_{\text{max}}t_3 \quad \text{or } t_3 = \frac{10S}{v_{\text{max}}}$$

$$v_{\text{av}} = \frac{\text{Total displacement}}{\text{Total time}} = \frac{S + 2S + 5S}{\frac{2S}{v_{\text{max}}} + \frac{2S}{v_{\text{max}}} + \frac{10S}{v_{\text{max}}}}$$

$$\frac{v_{\text{av}}}{v_{\text{max}}} = \frac{8S}{14S} = \frac{4}{7}$$

Alternative:

=WEEKLY TEST SOLUTION - TYJ =

$$\frac{v_{\text{av}}}{v_{\text{max}}} = \frac{\text{Total displacement}}{2\left(\frac{\text{Total displacement}}{\text{during acceleration}}\right) + \left(\frac{\text{Displacement}}{\text{during uniform}}\right)}$$

$$\frac{v_{\text{av}}}{v_{\text{max}}} = \frac{8S}{2(S+5S)+2S} = \frac{8}{14} = \frac{4}{7}$$

5. (b)
$$\sin \alpha = \frac{u}{v} = \frac{\sqrt{3}}{2} \implies \alpha = 60^{\circ}$$

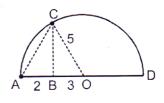
$$\Rightarrow \theta = 90^{\circ} + \alpha = 150^{\circ}$$

6. (a) For the person to be able to catch the ball, the horizontal component of velocity of the ball should be same as the speed of the person, i.e.,

$$v_0 \cos \theta = \frac{v_0}{2}$$
 or $\cos \theta = \frac{1}{2}$ or $\theta = 60^\circ$

8. From triangle BCO \Rightarrow BC = 4

From triangle BCA \Rightarrow AC = $\sqrt{2^2 + 4^2} = 2\sqrt{5}$ AC = $u_1 t_1$ BC = $u_2 t$ $\therefore \frac{u_1}{u_2} = \frac{AC}{BC} = \frac{2\sqrt{5}}{4} = \frac{\sqrt{5}}{\sqrt{4}}$



9. After 10 sec

$$\frac{u_{B}}{A} = 2 \times 10 = 20$$

$$A = \frac{1}{2} \times a \times 10^{2}$$

$$= 100$$

$$Now x_{A} = (40 \text{ t})$$

$$x_{B} = 100 + (ut) + \frac{1}{2}(2) t^{2} = 100 + 20 \text{ t} + t^{2}$$

6 km

Relative velocity,

A will be ahead of B when

$$x_B < x_A$$
 \Rightarrow 100 + 20 t + t² < 40 t
 \Rightarrow t² - 20 t + 100 < 0
 t^2 - 10t - 10 t + 100 < 0
 $t(t-10)$ - 10 (t - 10) < 0
 $(t-10)^2$ < 0

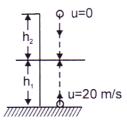
which is not possible

10. Height of the building

$$H = h_1 + h_2$$

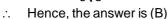
$$= \frac{1}{2}gt^2 + ut - \frac{1}{2}gt^2$$

$$= ut = 60 \text{ m}.$$



11. Velocity of rain = Velocity of man + Relative velocity of rain OR gives the actual velocity.

$$tan 30^{\circ} = \frac{VR}{OR}$$
$$= \frac{1}{\sqrt{3}} = \frac{6}{OR}$$
$$OR = 6\sqrt{3}$$



12.
$$t = \frac{AB}{\sqrt{5^2 - 3^2}} = \frac{3}{4} = 45 \text{ minutes}$$

:. Answer is (C)

13. Disance covered in 15 minutes =
$$5 \text{km/hr} \times \frac{15}{60} \text{hr} = 1.25 \text{ km}$$

Extra distance along river covered =
$$\sqrt{(1.25)^2 - (1)^2}$$
 = 0.75 km

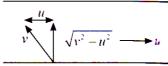
Velocity of river =
$$\frac{0.75}{(15/60)\text{hr}} = \frac{0.75 \times 4}{1} = 3\text{km/hr}$$

:. Answer is (B)

14. Let velocity of man in still water be v and that of water with respect to ground be u. Velocity of man downstream = v + u

As given,
$$\sqrt{v^2 - u^2} \ t = (v + u)T$$

 $\Rightarrow (v^2 - u^2)t^2 = (v + u)^2T^2$
 $\Rightarrow (v - u)^2 = (v + u)T^2$



[CHEMISTRY]

16. 34 electrons

17.

18. Bond orders are : $He_2^+ = 0.5$; $O_2^- = 1.5$; NO = 2.5; $C_2^{2-} = 3.0$

19.

- 20. XeF has 8 electrons in valence shell. In XeF₂, XeF₄ and XeF₆, two sigma bonds, four sigma bonds and six sigma bonds are respectively formed. Hence, in XeF₂ 3 pairs of electrons are left, in XeF₄ 2 pairs of electron are left and in XeF₆ only 1 pair of electron is left.
- 21. Each f C¹ and C² are forming two sigma bonds. Hence, both are sp-hybridised.
- 22. CO has triple bond $: \overline{C} = \overset{+}{\bigcirc} : CO_2$ has double bonds O=C=O,

 CO_3^{2-} has C–O bond intermediate between single and double bond.

- 23. In methane C-atom is sp³-hybridized with 25 s-character. In ethene, it is sp² with 33 s-character has to be less than 25 (actual value is 21.43)
- 24. Bond orders are : $O_2^- = 1.5$, NO = 2.5, $C_2^{2-} = 3.0$
- 25. $O = \stackrel{\oplus}{\underset{\alpha}{\mathbb{N}}} = O$ $\stackrel{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\beta}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}{\underset{\alpha}}}{\overset{\bullet}}{\overset{\bullet}}{\underset{\alpha}}}{\overset{\bullet}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}{\underset{\alpha}}}}{\overset{\bullet}}{\overset{\bullet}{\underset{\alpha}}{\overset{}}{\underset{\alpha}}}{\overset{\bullet}}{\overset{\bullet}{\underset{\alpha}}}{\overset{\bullet}}}}{\overset{\bullet}}{\underset{\alpha}}{\overset{\bullet}{\underset{\alpha}}}}{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}}}{\overset{\bullet}}{\underset{\alpha}}{\overset{\bullet}}}}}{\overset{\bullet}}{\overset{\bullet}}{\underset{\alpha}}}}}}{\overset{\bullet}}}{\overset{\bullet}}{\overset{\bullet}}{\underset{\alpha}}}{\overset{\bullet}}}{\overset{\bullet}}{\underset{\alpha}}}}}}}{\overset{\bullet}}{\overset{\bullet}}}}}}{\overset{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}}}}}{\overset{\overset{\bullet}}{\underset{\alpha}}}}}}}{\overset{\overset{\bullet}{\underset{\alpha}}}}}}{\overset{\overset{\bullet}}{\overset{\bullet}}{\underset{\alpha}}}}}}}{\overset{\overset{\bullet}}{\underset{\alpha}}}}}}}}{\overset{\overset{\bullet}}{\underset{\alpha}}{\overset{\bullet}}{\underset{\alpha}}}}}}}{\overset{\overset{\bullet}{\underset{\alpha}}}}}}}{\overset{\overset{\bullet}}{\underset{\alpha}}}}}}}{\overset{\overset{\bullet}{\underset{\alpha}}}}}}{\overset{\overset{\bullet}}{\underset{\alpha}}{\overset{\overset{\bullet}}}{\underset{\alpha}}}}}}}}{\overset{\overset{\overset{\bullet}{\underset{\alpha}}{\overset{\bullet}}}}}{\overset{\overset{\bullet}{\underset{\alpha}}}{\overset{\overset{\bullet}}}}{\underset{\alpha}}}}}}}}{\overset{\overset{\overset{\overset{\bullet}}}{\underset{\alpha}}}}}}}{\overset{\overset{\overset{\overset{\overset{}}}{\underset{$

26.

- 27. Bond order of N_2^{2-} and N_2^{2+} is 2.
 - Bond order of N_2^{2-} and N_2^{2+} is 2.5

Bond order of N₂ is 3

28. Bond orders of O_2^{2-}, O_2^{-}, O_2 and O_2^{+} are 1, 1.5, 2 and 2.5 respectively. (Please, refer to the text article no. 5.25)